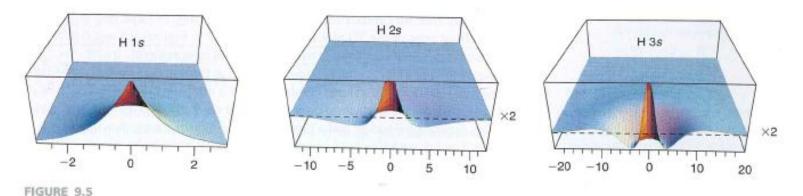


FIGURE 9.4

(a) 3D perspective and (b) contour plot of ψ₁₀₀(r). Red and blue contours correspond to the most positive and least positive values of the wave function, respectively. Colored version of figure on IX-12

Wavefunction

[1] Engel, Quantum Chemistry and Spectroscopy



Three-dimensional perspective plots of the 1s, 2s, and 3s orbitals. The dashed lines indicate the zero of amplitude for the wave functions. The " $\times 2$ " refers to the fact that the amplitude of the wave function has been multiplied by 2 to make the subsidiary maxima apparent. The horizontal axis shows radial distance in units of a_0 .

Colored version of figure on IX-13 [From [1]]

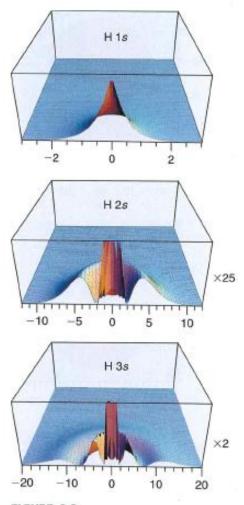
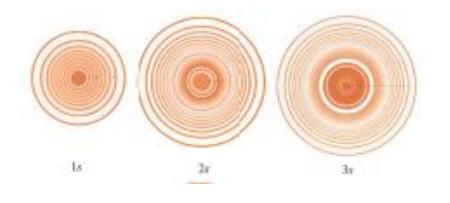


FIGURE 9.8

3D perspective plots of the square of the wave functions for the orbitals indicated. The numbers on the axes are in units of a_0 . The "×25" refers to the fact that the amplitude of the wave function has been multiplied by 25 to make the subsidiary maxima apparent.

Colored version of figure on IX-15 [From [1]]

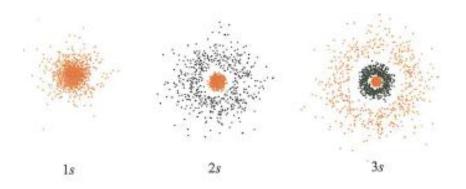
Wavefunction squared



Figures on IX-16

Wavefunction squared (probability density)
Contour plots

Using dots (density of dots is proportional to probility)



[2] McQuarrie, "Quantum Chemistry"

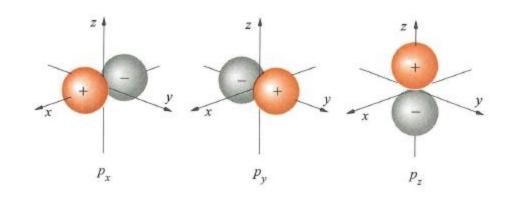


Figure on IX-22 [From [2]]
p orbitals (real angular functions)

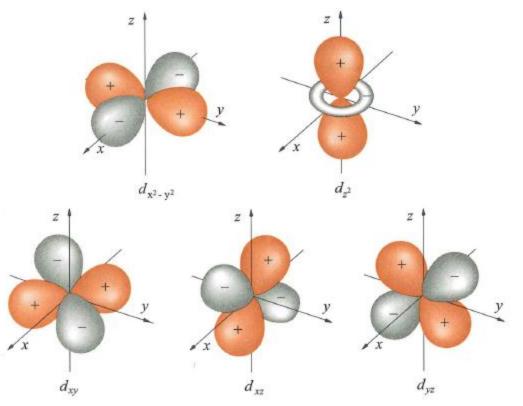
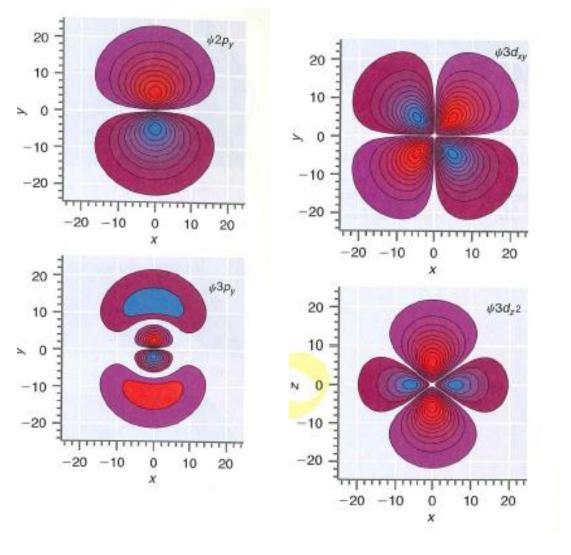


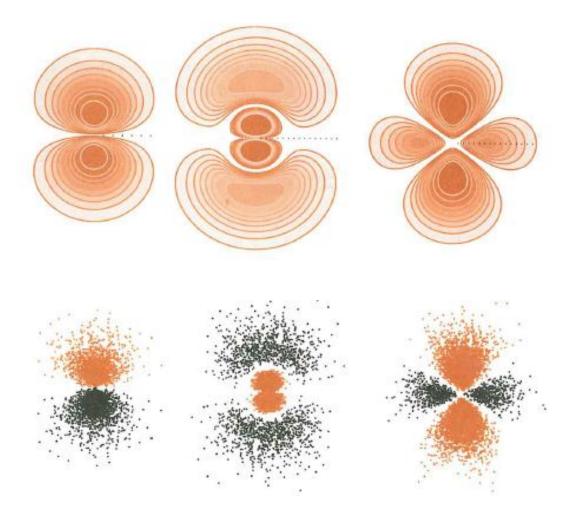
Figure on IX-24 [From [2]]

d orbitals (real angular functions)

Figures on IX-26 [From [1]]



Wavefunctions



Figures on IX-27 [From [2]]

Wavefunction squared (probability density) for (210), (310), (320)

P(r) = Radial probability distribution function [From [1]]

